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Quantum Chemical Investigations on the New Binary Anions $[GeSb_3]^{5-}$, $(Ge_2Sb_2)^{2-}$, $(Ge_4Sb_{12})^{4-}$, and $(Ge_4Sb_{14})^{4-}$

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Introduction

From its beginnings more than a hundred years ago to its developments in recent years, *Zintl* chemistry has become remarkably diverse.^[1–5] Our group has been actively involved in these endeavours.^[6–10] Among other methods, we use quantum chemical calculations to deepen our understanding of the geometric and electronic structures of binary Zintl anions.

we were able to afford a number of novel distinct anions, namely the carbonate-analogous [GeSb₃]⁵⁻ (1), the *pseudo*-tetrahedral anion $(Ge_2Sb_2)^{2-}$ (2), $(Ge_4Sb_{12})^{4-}$ (3), and $(Ge_4Sb_{14})^{4-}$ (4).^[11] One of these anions, $(Ge_2Sb_2)^{2-}$, was predicted to be synthetically accessible in one of our earlier works.^[12]

Starting from the new solid phase of the nominal composition "K₂GeSb",

Herein, we present in-depth results of the quantum chemical studies of these heretofore unknown Zintl anions.

Results

The starting mixture afforded the single-crystalline alloy $K_{12}Ge_{3.5}Sb_6$, which is comprised of a double-salt of " $K_2Ge_{1.5}$ " and $K_5[GeSb_3]$, with $[GeSb_3]^{5-}$ being a new member of a series of carbonate analogs. We furthermore obtained the heretofore unknown anions $(Ge_2Sb_2)^{2-}$ and $(Ge_4Sb_{12})^{4-}$ upon extraction with crypt-222 and en. Addition of [AuMe(PPh₃)] yielded the anion (Ge₄Sb₁₄)⁴⁻, the lighter homolog of the already known (Ge₄Bi₁₄)⁴⁻.^[8] Synthetic details are given by Katrin Beuthert at **Poster 12**.







Bond Lengths / A		
	Calculated	Experiment
Ge–Ge	2.55	2.5316
Ge–Sb	2.73	2.6941 – 2.7057
Sb–Sb	2.83	2.7889





Eight isomers are found to be within 10 kJ/mol with respect to the global minimum structure.

+2 kJ/mol



Only 2e2c bonds







0 kJ/mol

+4 kJ/mol

+4 kJ/mol





+7 kJ/mol



+3 kJ/mol



The ratio of the covalent radii

(Q_{cr}) determines the distribution

of the different atom types

within the $(Ge_4Pn_{14})^{4-}$ anions

(Pn: P, As, Sb, Bi).

0 kJ/mol





+5 kJ/mol





+168 kJ/mol





0 kJ/mol

0 kJ/mol

+152 kJ/mol



+13 kJ/mol





+9 kJ/mol

+10 kJ/mol +9 kJ/mol

0 kJ/mol +12 kJ/mol

+135 kJ/mol

Conclusion

- Four novel binary anions were obtained from " K_2 GeSb".
- DFT studies helped to elucidate their geometric and electronic structures.
- A mixture of different isomers of **3** is likely to be present in solution.
- The distribution of different atom types over the respective sites is heavily influenced by the ratio of the ovalent radii (Q_{cr}) .

Outlook

- Further investigations of the reactions of binary *Zintl* anions with transition metal complexes are currently underway.
- They are being accompanied by extensive quantum chemical studies.

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