

Eyesight to AI: Discovery of effective corrosion modulators via predictive machine learning models

T. Würger, C. Song, B. Vaghefinazari, A. Lisitsyna, P. Fischer, G. Wiese, M. Zheludkevich, S. Albarqouni, S. V. Lamaka, C. Feiler

Helmholtz-Zentrum Hereon

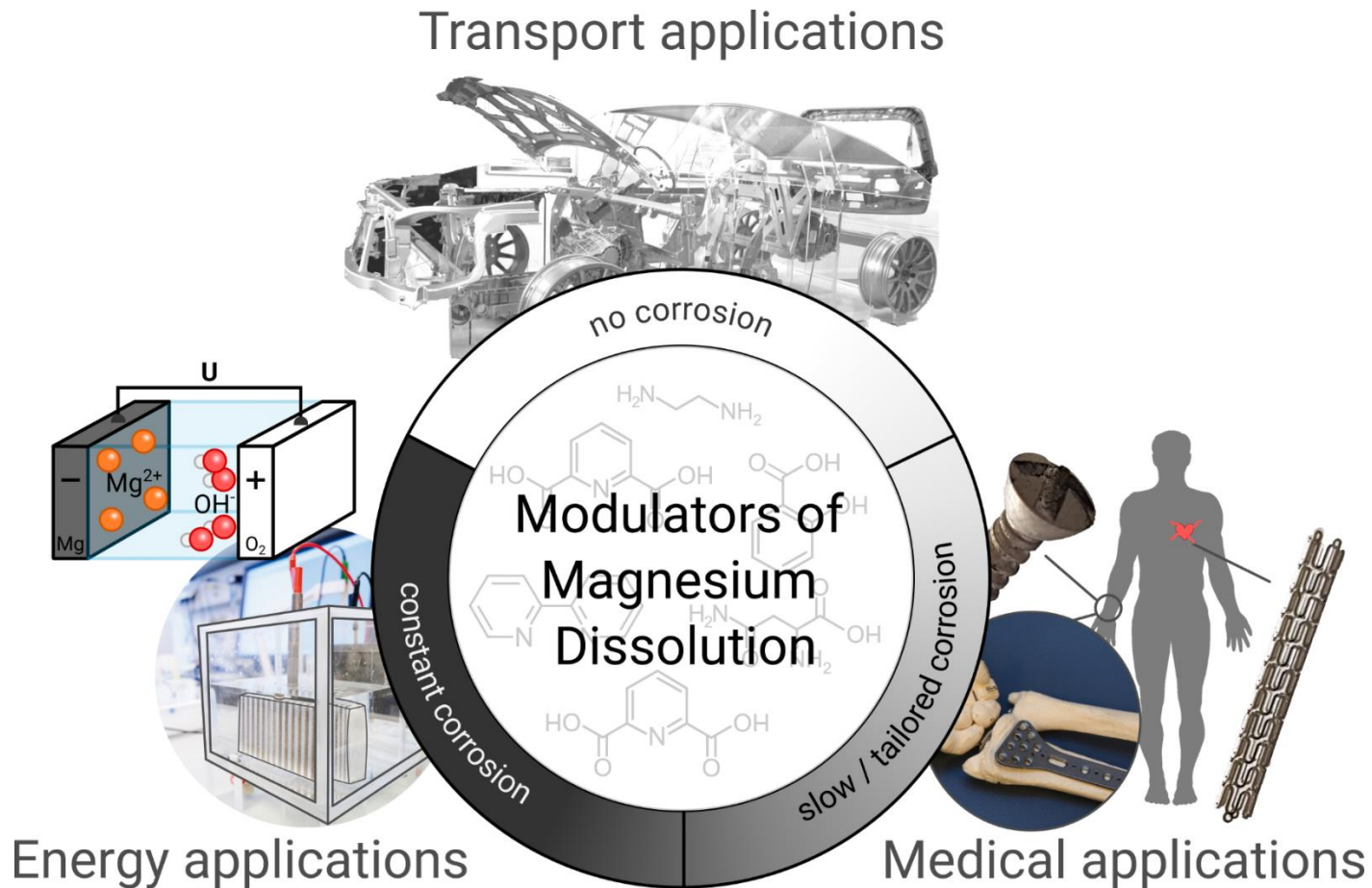
MSE Day

14.11.2023



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Dissolution Modulators



- Magnesium is the lightest structural metal
- Highly versatile engineering material
- Degradation control is vital for application
- Small organic molecules can be employed as dissolution modulators
 - Chemical Abstracts Service database: $1.5 \cdot 10^8$
 - Estimated number of synthesizable compounds: $1 \cdot 10^{63}$
- Data-driven approaches depict great tools to screen this vast chemical space

Images by courtesy of S. V. Lamaka, C. Blawert, B. Vaghefinazari and T. Würger.
A. Mullard, *Nature* **2017**, 549, 445-447.

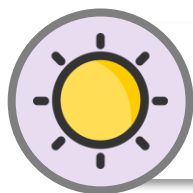
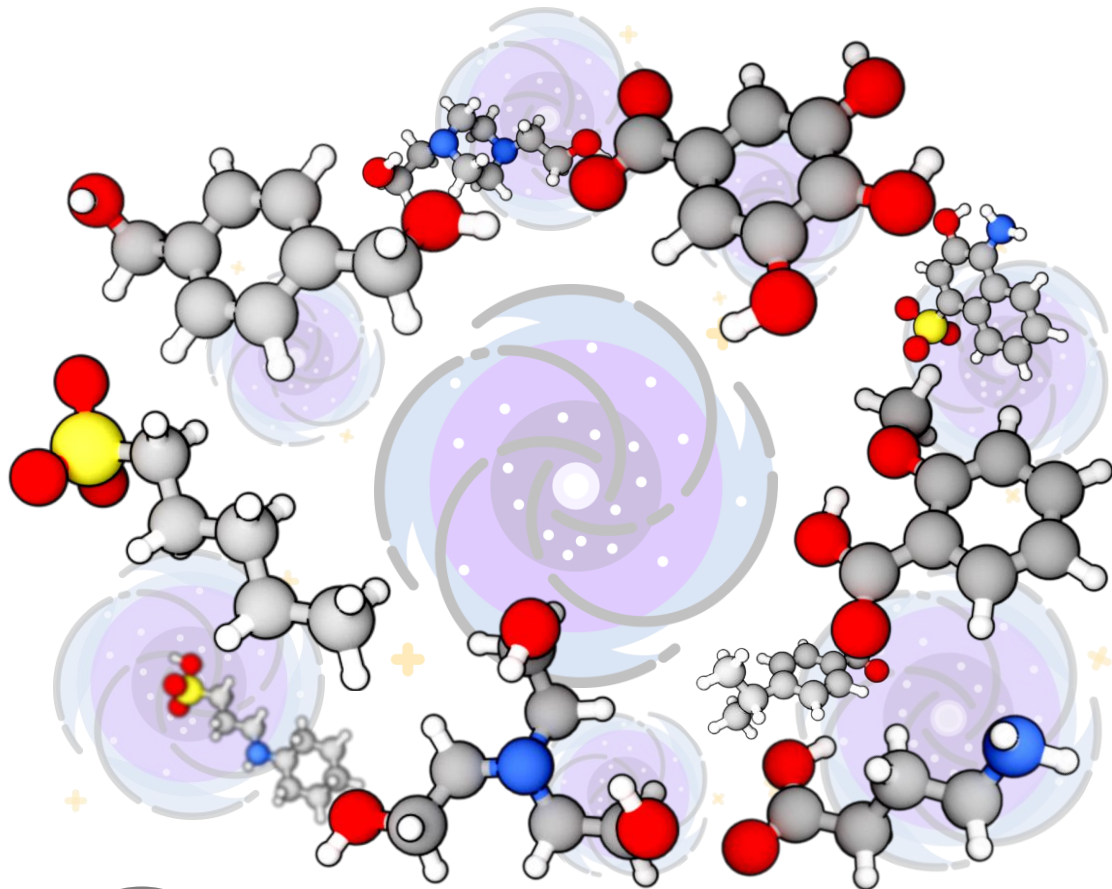
D. A. Erlanson, S. W. Fesik, R. E. Hubbard, W. Jahnke, H. Jhoti, *Nature Reviews Drug Discovery* **2016**, 15, 605-619.

D. Höche, S. V. Lamaka, B. Vaghefinazari, T. Braun, R. Petrauskas, M. Fichtner, M.L. Zheludkevich, *Scientific Reports* **2018**, 8, 7578.

S. V. Lamaka, B. Vaghefinazari, D. Mei, R. P. Petrauskas, D. Höche, M. L. Zheludkevich, *Corrosion Science* **2017**, 128, 224-240.

Exploring the Chemical Space

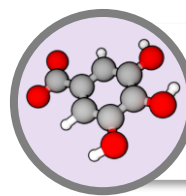
Thought Experiment: “Just” a Matter of Time



Approximately $1.0 \cdot 10^{21}$ stars in the observable universe

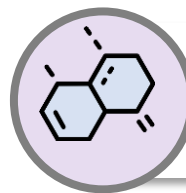


- Automated setup to quantify influence on corrosion rate
- Ten experiments in parallel (duration 24 h)



Chemical Abstracts Service database lists $1.5 \cdot 10^8$ organic chemicals

~ 41095 years

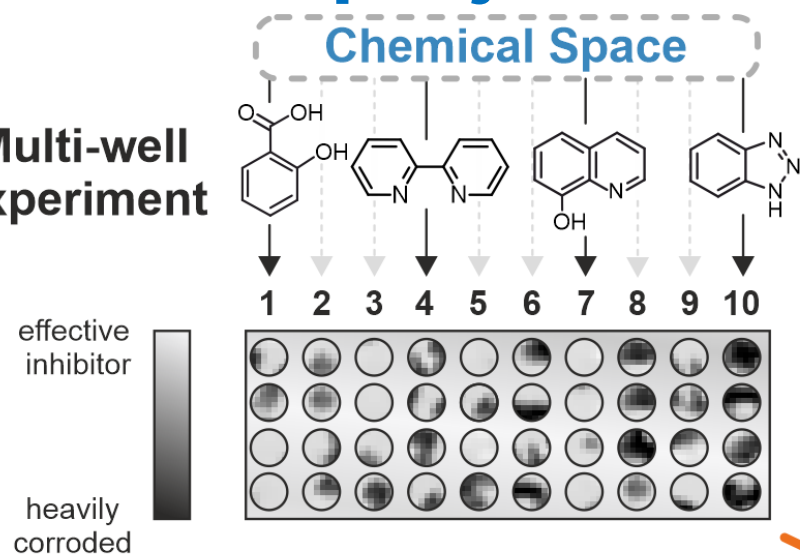


Estimated number of synthesisable compounds amounts to $1.0 \cdot 10^{63}$

~ $2.73 \cdot 10^{59}$ years

The AI² project

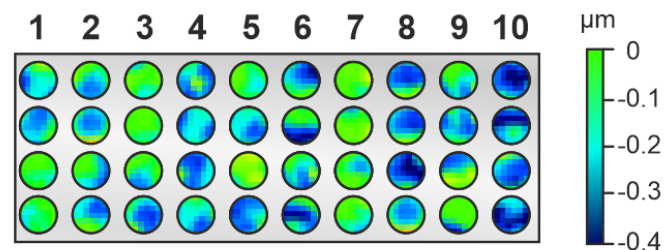
Multi-well experiment



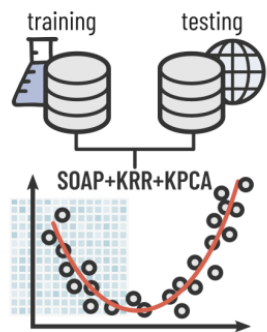
Topography assessment & Image acquisition

→ Phase 1
→ Phase 2

Ground truth generation



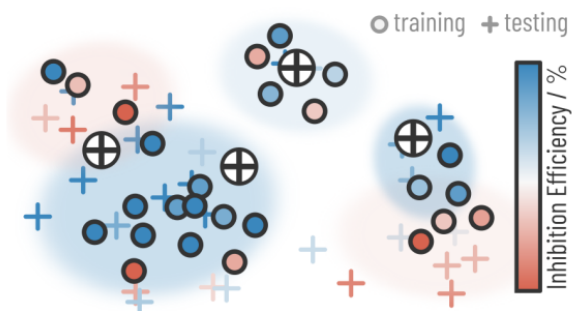
Prediction of untested compounds



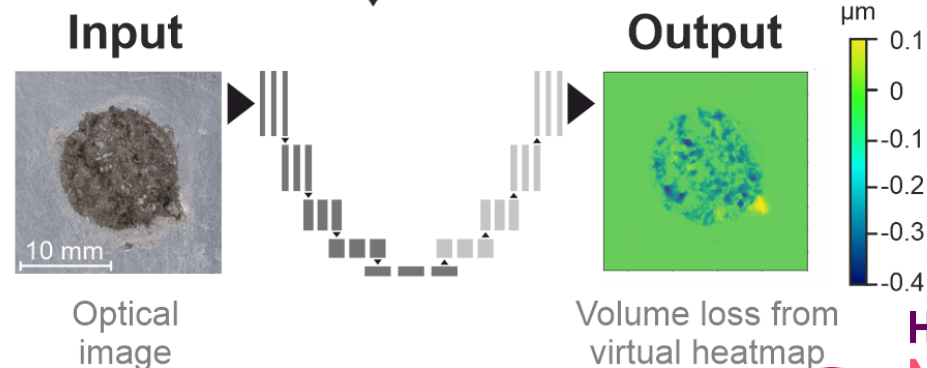
Generation of Quantitative Structure-Property Relationship models

image acquisition

Training database for QSPR



Training data for image recognition



UNet-based Image-to-Image translation



Sviatlana Lamaka



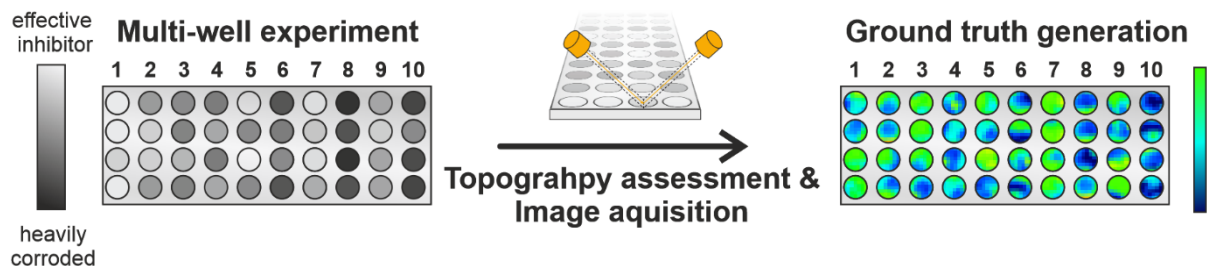
Shadi Albarqouni

- The project team asked me to remove all the technical content from the slides as we are on the verge of submitting it to a peer-reviewed journal
- However, we are happy to discuss in more detail in personal meetings and are also very open to potential collaborations!

Summary

Mirror mirror on the wall who is the best inhibitor of them all?

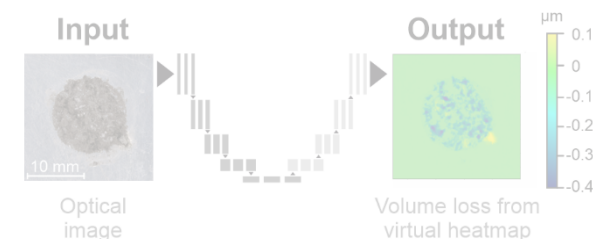
Determination of weight loss from optical appearance of corrosion imprints



- Data generation using a multi-well setup
- Generation of heatmaps and determination of volume loss in the well by profilometer
- 200 distinct compounds have been tested

- Determination of volume loss based on optical appearance
- Eliminates need to perform manual (time-consuming) analysis of imprints
- Correlation between volume loss from experiment and virtual heatmap
- Highly sensitive to minimal offsets during annotation step

UNet-based generation of virtual heatmaps



Performance prediction of untested chemicals by QSPR

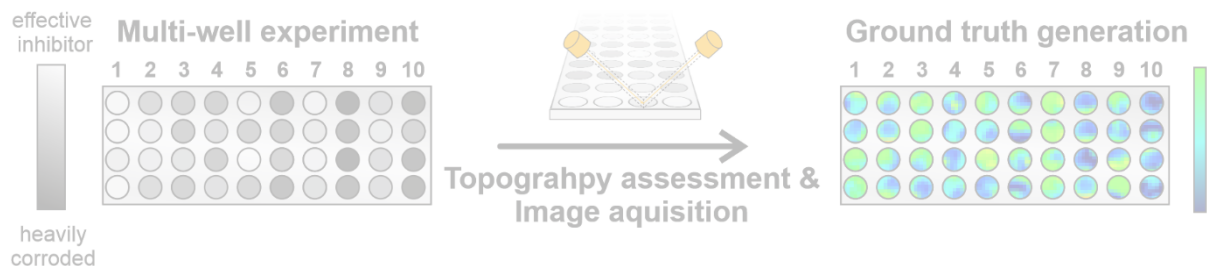


- Training of SOAP-based QSPR model based on ground truth
- Use of additional factors for compound selection (toxicity, solubility, price, uncertainty) rather than only based on molecular similarities
- First blind testing conducted, Next step: active learning and use of volume loss derived from virtual heatmaps as target for training of QSPR models

Summary

Mirror mirror on the wall who is the best inhibitor of them all?

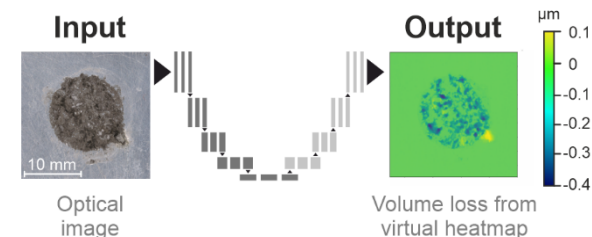
Determination of weight loss from optical appearance of corrosion imprints



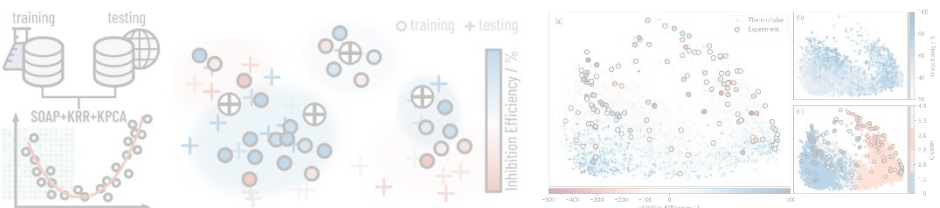
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UNet-based generation of virtual heatmaps



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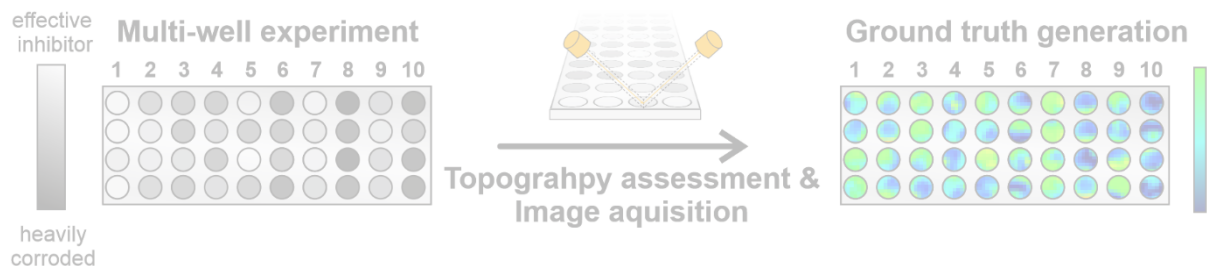


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Determination of weight loss from optical appearance of corrosion imprints



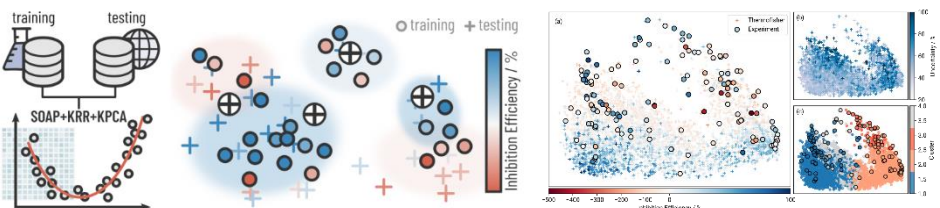
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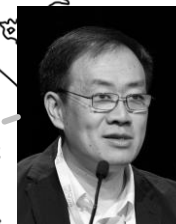


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Inhibitor Database ExCorr



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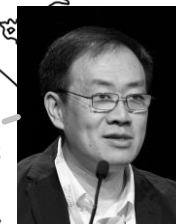


ExCorr

Explore the Chemical Space of
Corrosion Inhibitors

[Go to database](#) [About](#)

Inhibitor Database ExCorr



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Search database		Filter database	
IUPAC	CAS	SMILES	MW / Da
quinolin-8-ol (8-hydroxyquin)	148-24-3	<chem>O=C2N=CC=CC2=C1</chem>	145.16
quinolin-8-ol(8-hydroxyquir	148-24-3	<chem>O=C2N=CC=CC2=C1</chem>	145.16
quinoline-2-carboxylic acid (l	93-10-7	<chem>O=C(O)C1=NC2=CC=CC=C2</chem>	173.17

Base	Alloy	Method	IE / %	pH	Aggressive Component	Concentration / mM	References
Al	AA2024-T3		75		0.05M NaCl		(Lamaka et al.;2007)Balaskas et
Al	AA2024-T3		94		0.05M NaCl		(Lamaka et al.;2007)Balaskas et
Al	AA2024-T3	impedance	89.8		3.5% NaCl	5	(Lamaka et al.;2007)Balaskas et

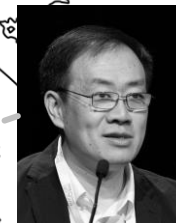
Show similar structures

IUPAC	CAS	SMILES	MW / Da
2-phenylchromen-4-one (2-f	525-82-6	<chem>O=C1C=C(C2=CC=CC=C2)O</chem>	222.24
2H-Chromen-2-one(1;2-Ben	91-64-5	<chem>C1=CC=C2C(=C1)C=CC(=O</chem>	146.14
3';6'-dihydroxyspiro[2-benzo	2321-07-5	<chem>O=C1OC2(C3=C(C(OC4=C2C=</chem>	332.31

Base	Alloy	Method	IE / %	pH	Aggressive Component	Concentration / mM	References
Al	powder	H2 evolution	45		1M HCl, 2.5% NaCl	10	(Horner and Meisel;1978)



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Search database

IUPAC CAS

quinolin-8-ol (8-hydroxyquin) 148-24-3

quinolin-8-ol(8-hydroxyquin) 148-24-3

quinoline-2-carboxylic acid (l) 93-10-7

Base	Alloy	Method	IE / %	pH	Aggre
Al	AA2024-T3		75		0.05M
Al	AA2024-T3		94		0.05M
Al	AA2024-T3	impedance	89.8	3.5%	

2-phenylchromen-4-one (2-phenylchromone) 525-82-6

2H-Chromen-2-one(1;2-Ben) 91-64-5

3',6'-dihydroxyspiro[2-benzo] 2321-07-5

Base	Alloy	Method	IE / %	pH	Aggre
Al	powder	H2 evolution	45		1M

Filter options

Base Material

Mg X Al X Cu X Fe X

Alloy

CPMg220ppmFe X HPMg51ppmFe X HPMg50ppmFe X

WE43 X ZE41 X E21 X AZ31 X AZ91 X AM50 X

pure X AA1060 X AA1100 X AA2024 X AA2024-T3 X

AA2056 X AA2198 X AA3003 X AA5052 X AA5754 X

AA6061 X AA6063 X AA6082 X AA7075 X

AA7075-T6 X Al20556 X AlSi11MgSr X CPAL X

pigment X powder X CPMg342ppmF X CU10Zn X

CU40Zn X AS1204-250 X carbon steel X mild steel X

AlSi1020 X steel X

Method

H2 Evolution X Potentiodynamic Polarization X Impedance X

Weight Loss X Gravimetry X Thermometric X

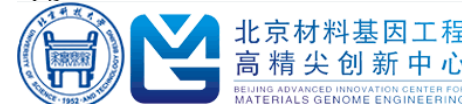
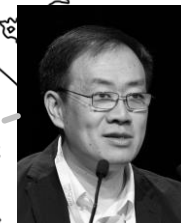
Adsorption Spectroscopy X Optical Inspection X

pH

Minimum Inhibition Efficiency / %

50

Inhibitor Database ExCorr



ExCorr

- Review contains experimental information of 2400 compounds for Al-, Cu-, Fe- and Mg-based materials tested at various operating concentrations and with different techniques
- >1000 are already in a machine-readable format (roughly 3000 entries)
- Experimentalists can submit new data via a mask and ExCorr will be updated on a regular basis
- Will be released together with the review.
- <https://excorr.web.app/>

Filter options

Base Material

Mg × Al × Cu × Fe ×

Alloy

CPMg220ppmFe × HPMg51ppmFe × HPMg50ppmFe ×
WE43 × ZE41 × E21 × AZ31 × AZ91 × AM50 ×
pure × AA1060 × AA1100 × AA2024 × AA2024-T3 ×
AA2056 × AA2198 × AA3003 × AA5052 × AA5754 ×
AA6061 × AA6063 × AA6082 × AA7075 ×
AA7075-T6 × Al20556 × AlSi11MgSr × CPAL ×
pigment × powder × CPMg342ppmF × CU10Zn ×
CU40Zn × AS1204-250 × carbon steel × mild steel ×
AlSi1020 × steel ×

Method

H2 Evolution × Potentiodynamic Polarization × Impedance ×
Weight Loss × Gravimetry × Thermometric ×
Adsorption Spectroscopy × Optical Inspection ×

pH

-1 0 1 2 3 4 5 6 7 8 9 10 11 12 13 14

Minimum Inhibition Efficiency / %

50

Thank you!

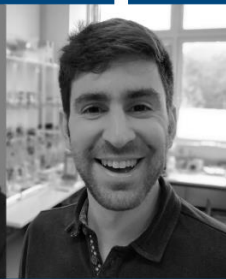
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RESEARCH FOR GRAND CHALLENGES



Dr. Sviatlana
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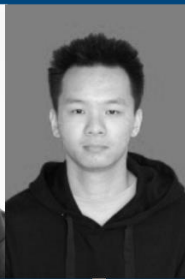
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Dr. Tim
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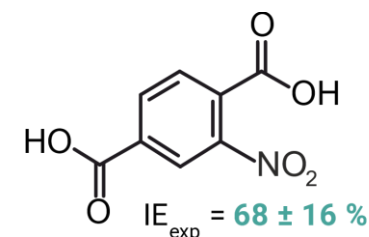
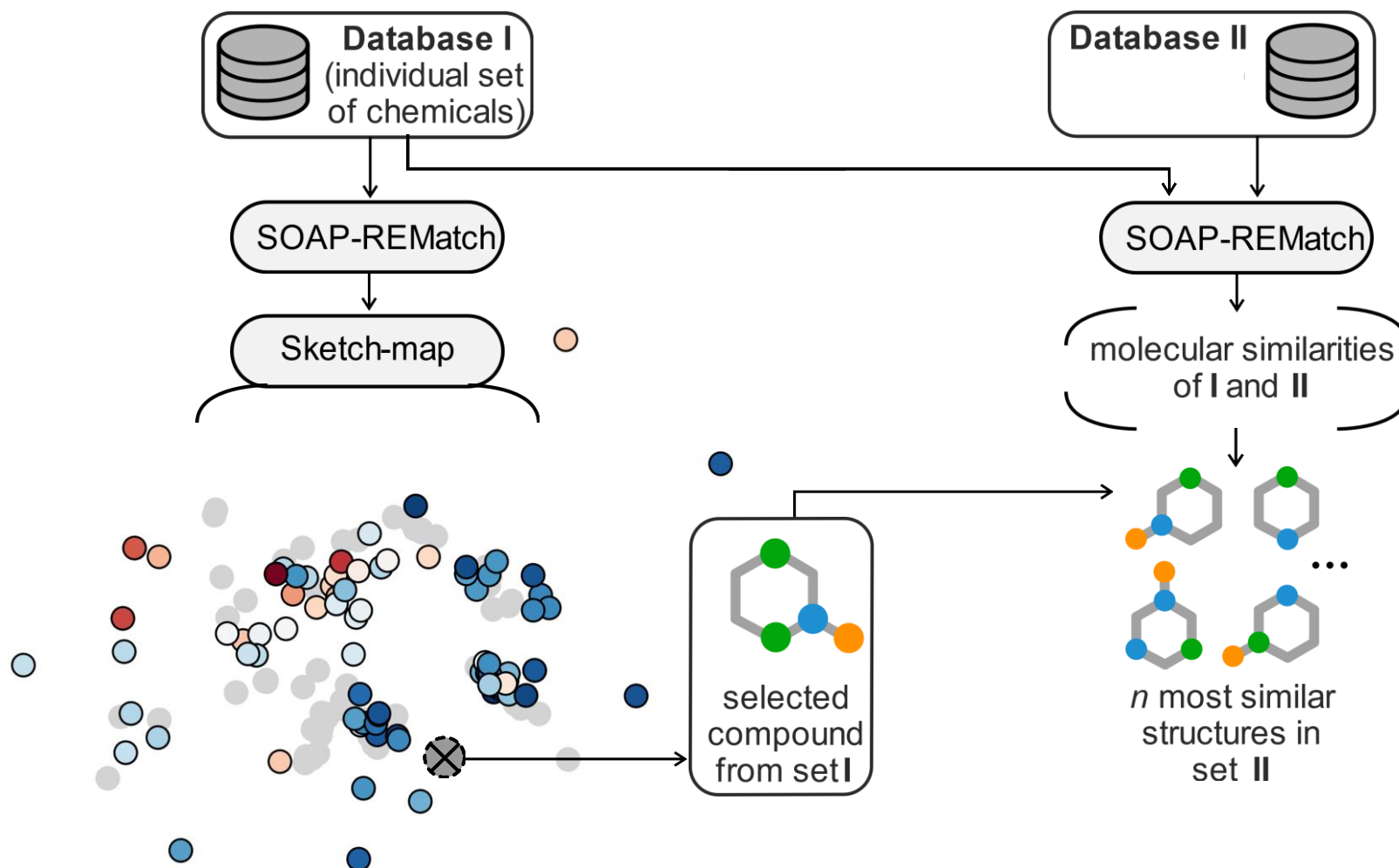
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ExChem

Similarity-Based Selection of Compounds of Interest



Selected inhibitor from individual data set



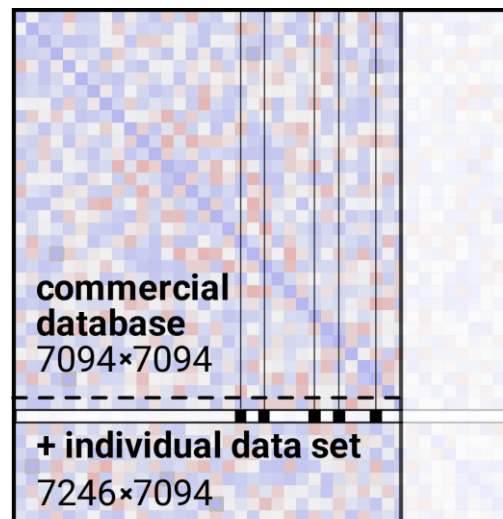
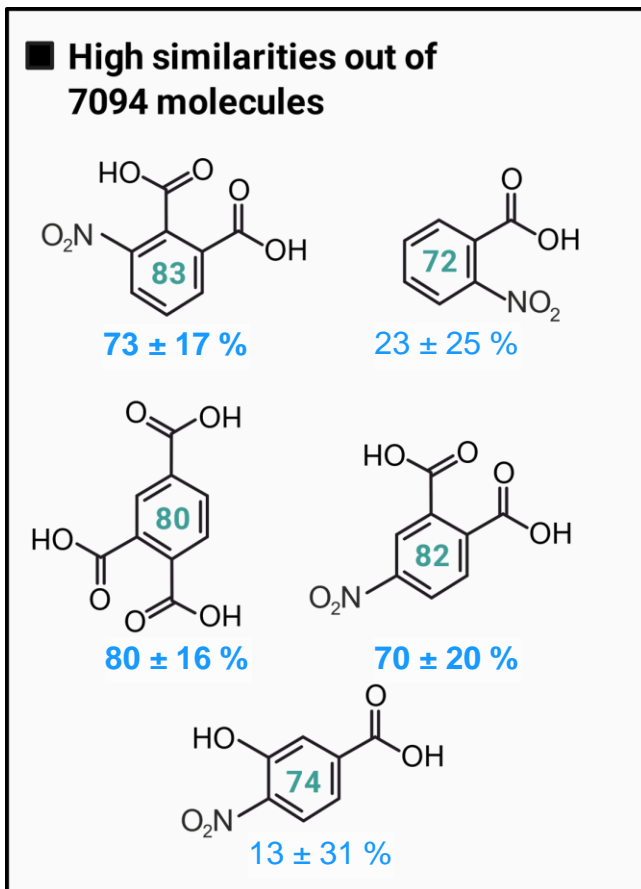
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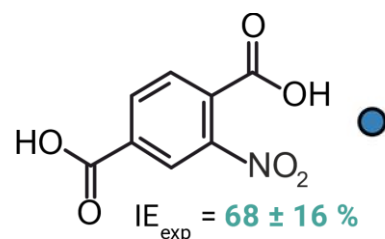


ExChem

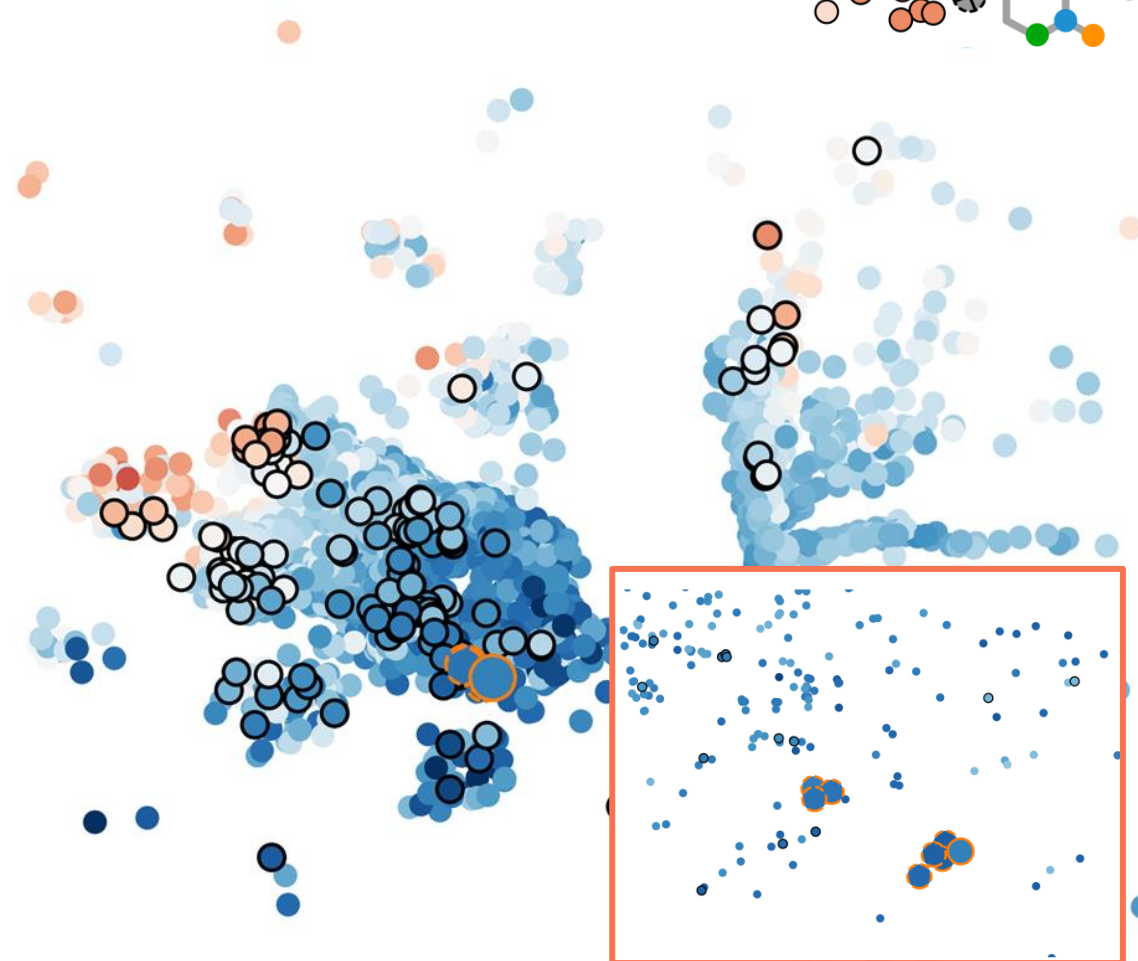
Similarity-Based Selection of Compounds of Interest



Select index in **global similarity matrix** (7246×7246)



Selected inhibitor from individual data set



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Exploring structure-property relationships in magnesium dissolution modulators.
npj Mater Degrad 5, 2 (2021). <https://doi.org/10.1038/s41529-020-00148-z>

